

Give the *IUPAC* name for each of the following:

$\text{Mn}(\text{ClO}_3)_2$	manganese (II) chlorate	ClF_3	chlorine trifluoride
$\text{Mn}(\text{CO}_3)_2$	manganese (IV) carbonate	XeF_4	xenon tetrafluoride
PCl_3	phosphorus trichloride	FeI_3	iron (III) iodide
$\text{H}_2\text{CO}_3(\text{aq})$	carbonic acid	$\text{Fe}(\text{ClO}_4)_2$	iron (II) perchlorate
V_2O_5	vanadium (V) oxide	Ag_3PO_4	silver phosphate
$\text{Ba}(\text{IO}_2)_2$	barium iodite	Ca_3P_2	calcium phosphide
$\text{Al}_2(\text{SO}_3)_3$	aluminum sulfite	$\text{HI}(\text{aq})$	hydroiodic acid
$\text{Na}_2\text{SO}_4 \cdot 4\text{H}_2\text{O}$	sodium sulfate tetrahydrate	CuSO_4	copper (II) sulfate
PbCl_4	lead (IV) chloride	$\text{Cd}(\text{C}_2\text{H}_3\text{O}_2)_2$	cadmium acetate
CuMnO_4	copper (I) permanganate	N_2O_4	dinitrogen tetroxide

Give the chemical formula for each of the following:

selenium dioxide	SeO_2	sulfur trioxide	SO_3
lead (IV) sulfate	$\text{Pb}(\text{SO}_4)_2$	aluminum hydroxide	$\text{Al}(\text{OH})_3$
lead (II) sulfite	PbSO_3	titanium (IV) oxide	TiO_2
potassium sulfide	K_2S	scandium (II) phosphate	$\text{Sc}_3(\text{PO}_4)_2$
chlorous acid	$\text{HClO}_2(\text{aq})$	boron trifluoride	BF_3
cobalt (II) nitride	Co_3N_2	carbon monoxide	CO
bromine triiodide	BrI_3	vanadium (II) iodate	$\text{V}(\text{IO}_3)_2$
cesium hypobromite	CsBrO	silicon tetrachloride	SiCl_4
carbon tetrachloride	CCl_4	hydrocyanic acid	$\text{HCN}(\text{aq})$
carbon disulfide	CS_2	manganese (IV) chlorite	$\text{Mn}(\text{ClO}_2)_4$
aluminum phosphate	AlPO_4	bromine trifluoride	BrF_3